

Science

Key Stage 2 Curriculum

During years 5 and 6, pupils should be taught to use the following practical scientific methods, processes and skills through the teaching of the programme of study content:

- planning different types of scientific enquiries to answer questions, including recognising and controlling variables where necessary
- taking measurements, using a range of scientific equipment, with increasing accuracy and precision, taking repeat readings when appropriate recording data and results of increasing complexity using scientific diagrams and labels, classification keys, tables, scatter graphs, bar and line graphs
- using test results to make predictions to set up further comparative and fair tests
- reporting and presenting findings from enquiries, including conclusions, causal relationships and explanations of and degree of trust in results, in oral and written forms such as displays and other presentations
- identifying scientific evidence that has been used to support or refute ideas or arguments.
- Common topics include: plants, animals (including humans), living things and their habitat, properties and changes of materials rocks, light, sound, electricity, space and forces.



	Year 7	Year 8	Year 9	Year 10	Year 11	Year 12	Year 13
Autumn 1	Introduction to science • Lab safety • Using Bunsen burners Block 1 Cells • Microscopes • Animal Cells • Plant cells • Specialised cells • Organisation Particles • Solids, liquids & gases • Properties of solids, liquids and gas • Changing state • Diffusion • Dissolving	Block 1 Digestion • Food groups • Food tests • Digestive system organs • Journey of a sandwich • Disease caused by poor diet Metals and reactivity • Properties of metal • Oxides • Displacement reactions • Alkali metals	See, Biology, Chemistry, Physics and Applied Science				

	Gas pressure					
Autumn 2	Energy 1 - Energy stores - Energy transfers - Sankey diagrams - Efficiency Block 2 Reproduction - Male reproductive organs - Female reproductive organs - Fertilisation - Pregnancy - Puberty - Menstrual cycle - Plant reproduction Separating mixtures - Pure and impure - Filtration - Crystallisation - Distillation - Chromatography	Magnets and electromagnets - Permanent magnets - Magnetic fields - Electromagnets Block 2 Inheritance - Inheritance - DNA - Fertilisation - Punnet squares Types of reaction - Chemical and physical reactions - Conservation of mass - Combustion - Decomposition - Endo and exo				
Spring 1	Forces 1 - Speed - D-T graphs - Force arrows - Friction - Drag - Stretching - Forces and motion	Energy 2 - Work done - Power - Domestic energy use - Energy demand Block 3 Movement - Skeleton - Muscles				

	Block 3 Respiration and breathing • Aerobic respiration • Lungs • Anaerobic respiration • Smoking and asthma •	• Antagonistic pairs						
Spring 2	Acids and Alkalis • pH scale • neutralisation Electricity • Charge • Conductors and insulators • Current • PD • Resistance • Series • Parallel	Materials and the Earth • Structure of the Earth • Igneous • Sedimentary • Metamorphic • Rock cycle • Atmosphere • Climate change Forces 2 • Gravity and weight • Pressure • Floating and sinking • Levers • Moments						
Summer 1	Block 4 Photosynthesis • Structure of a plant • Plant leaf • Photosynthesis • Factors affecting photosynthesis Elements • Elements • Compounds • Mixtures • Reactions	Block 4 Infection • Pathogens and disease • Immunity • Vaccination Heating and cooling • Conduction • Convection • Radiation • Insulation						

	• Periodic Table • Equations							
Summer2	Universe • Planets • Galaxy • Universe • Big Bang • Light • Reflection • Refraction • Colour • The eye	Ecology • Keys • Food chains • Food webs • Classification • Sampling techniques • Pooters • Quadrats and line transits Sound • What is sound • Amplitude • Frequency • Waves • Echoes						

Biology

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Autumn 1	See KS3 Science		Cells • A typical animal cell • Plant cells • Prokaryotic and Eukaryotic cells • A typical Bacteria cell • The size of cells • Using microscopes to look at cells • Calculating magnification • Chromosomes • Mitosis and the cell cycle • Stem cells	Infection • Pathogens and disease • Viral diseases • Bacterial diseases • Protists and diseases • Fungal diseases • Preventing entry of pathogens • The immune system • Boosting immunity • Antibiotics • Developing new drugs	Inheritance • Asexual reproduction • Sexual reproduction and meiosis • The genome • Genetic inheritance • Genetic crosses • Inherited disorders • Sex determination • Variation • Evolution • Evidence for evolution • Selective breeding • Genetic engineering • Principles of classification • Extinction • Evolution trees	Biological molecules Structure and function of: • Carbohydrates • Lipids • Proteins • Nucleic acids • Water	Energy transfer in and between organisms • Photosynthesis • Respiration
Autumn 2				Bioenergetics • Photosynthesis • Factors affecting Photosynthesis • Converting glucose • The importance of respiration • Aerobic respiration		Cells • Structure and function of different types of cells • Mitosis and meiosis • Structure and function of membranes	Responding to internal and external environments • Stimuli and responses • Nervous control • Homeostasis
Spring 1			• Use of stem cells • Diffusion	• Anaerobic respiration	Ecology • Relationships between organisms	Exchange and transport	

Spring 2			- Factors affecting diffusion - Osmosis - Active transport - Comparing processes Organisation - Specialised cells - Tissues, organs and systems - Enzymes - Enzymes in digestion	- Exercise and respiration - Metabolism Homeostasis - The importance of hormones - Control systems - The nervous system - Endocrine system	- Adaptations - Studying ecosystems - Recycling materials - Feeding relationships - Biodiversity - Pollution - Overexploitation - Conservation biodiversity	- Gas exchange - Digestion - Cardiovascular system - Mass transport in plants including transpiration and translocation Genetic information, variation and relationships - Structure, function and replication of DNA - Protein synthesis - Genetic diversity and adaptations - Classification	Genetics, populations and evolution - Genetic diagrams - Variation, selection and speciation - Succession - Conservation Control of gene expression - Mutations - Stem cells - Gene expression - Genome projects and technologies
Summer 1			- Bile and digestion - Blood - Blood vessels - The heart - Gaseous exchange - Health and disease - Risk factors - Diseases of the heart - Cancer - Plant tissues - Water transport - Translocation	- Control of blood glucose - The sex hormones - Control of the menstrual cycles - Reducing fertility - Increasing fertility			
Summer 2							

Chemistry

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Autumn 1	See KS3 Science		Atoms • Atoms elements and compounds • Equations • Separating mixtures • Scientific models of the atom • Subatomic particles • Isotopes and ions • Electronic configuration	Energy and Chemical Change • Oxidation and reduction • The reactivity series • Displacement reactions • Extraction of metals • Losing and gaining electrons • The pH scale	Chemical analysis • Pure and impure substances • Formulations • Chromatography • Gas tests	Atomic structure • The atom • Atomic models • Relative mass • The mass spectrometer • Using mass spectra • Electronic structure • Ionising energies Amount of substances • The mole	Electrode potentials and cell • Electrode potentials • Standard electrode potentials • Electrochemical series • Electrochemical cells Acids, Bases and pH • Acids bases and Kw • pH calculations

Autumn 2		• The development of the periodic table • Group 0 • Group 7 • Group 1	• Neutralisation of Acids • Soluble salts and insoluble bases • Strong and weak acids • Electrolysis • Oxidation and reduction • Extraction of metals • Electrolysis of aqueous solutions		• Gases and the mole • Chemical equations • Titrations • Formulas • Chemical yield • Atom economy	• The acid dissociation constant • Titration d pH curves • Titration calculations • Buffer action • Calculating the pH of Buffers
				Atmosphere • The earths atmosphere • The atmosphere today • Increase of oxygen levels • Decrease of carbon dioxide levels • Greenhouse gases • The impact of human activity • Global climate change • Carbon footprints	Bonding • Ionic Bonding • Covalent Bonding • Chare clouds • Shapes of molecules • Polarisation • Intermolecular forces • Metallic bonding • Properties of metals Energetics • Enthalpy • Bond Enthalpies • Measuring Enthalpy changes • Hess' Law	Transition Metals • Transition metals • Complex ions • Isomerism in complex ions • Formation of coloured compounds • Ligand substitution reactions • Variable oxidation states • Transition metals titrations • Metal aqua ions Isomers and carbonyl compounds • Optical isomerism • Aldehydes ad Ketones • Hydroxynitriles • Carbocyclic acids and esters • Reaction's and uses of esters

						● Acyl chlorides ● Acid anhydrides ● Purifying organic compounds Aromatic compounds and amines ● Aromatic compounds ● Reactions of aromatics ● Amines and amides ● Reactions of amines
Spring 1		Bonding ● The three states of matter ● Changing states ● Identifying states of a substance ● State symbols ● Chemical bonds ● Ionic bonding ● Properties of ionic compounds ● Covalent bonding ● Small molecules ● Giant covalent structures ● Graphene ● Fullerenes ● Polymers ● Metallic bonding ● Properties of metals ● alloys	Rates ● Calculating the rate of reaction ● Collision theory ● Plotting the reaction rates ● Catalysts ● Reversible reactions ● Closed systems ● Changing reaction conditions	Using resources ● Sustainable development ● Drinking water ● Waste water treatment ● Alternative methods of extracting metals ● Life cycle assessment s ● Reducing the use of resources	Kinetics ● Reaction rates ● Catalysts ● Measuring reaction rates Equilibria and redox reactions ● Reversible reactions ● Industrial processes ● The equilibrium constant ● Factors effecting the equilibrium constant ● Redox reactions ● Redox equations	Polymers ● Condensation polymerisation ● Monomers and repeating units ● Disposing of polymers Amino acids, proteins and DNA ● Amino acids ● Proteins ● Enzymes DNA ● Cisplatin
Spring 2					Periodicity ● The periodic table ● Periodicity Group 2 and group 7 elements ● Group 2	Further synthesis and Analysis ● Organic synthesis ● NMR spectroscopy ● C NMR ● H NMR ● Chromatography

					• Group 2 compounds • Group 7 • Halide ions • Tests for ions Introduction to organic chemistry • Formulas • Functional groups • Nomenclature • Mechanisms] • Isomers • E/Z isomers	• Gas chromatography
Summer 1		Quantitate • Conservation of mass • Relative formula mass • Apparent change in mass • Amount of substances • Calculating the amount of substances • Balanced equations • Limiting reactions • Concentrations of solutions			Alkanes and Halogenalkanes • /alkanes and petroleum • Alkanes as fuels • Synthesis of chloroalkanes • Halogenalkanes • Nucleophilic substitution • Elimination reactions • Alkenes and alcohols • Alkenes • Reactions of alkenes • Additional polymers • Alcohols • Dehydration of alcohols • Ethanol production	

Summer 2					• Oxidation of alcohols • Organic Analysis • Tests for functional groups • Mass spectrometry • Infrared spectroscopy	
			Organics • Crude oils and hydrocarbons • Fractional distillation • Alkanes • Burning fuels • Cracking hydrocarbons • Bromine water		Thermodynamics • Enthalpy changes • Born Haber cycles • Enthalpies of solution • Entropy • Free energy change Rate equation and Kp • Monitoring reactions • Reaction rates and graphs • Rate equations • The initial rate methods • Clock reactions • Rate and concentration graph • The rate determining step • The Arrhenius equation • Gas equilibria • Changing gas equilibria	

Physics

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Autumn 1	See KS3 science		Particles • States of matter • Density • Specific Latent Heat • Specific heat Capacity • Particle motion • Pressure	Atomic Structure • The structure of the atom • Isotopes • The plum pudding model • Rutherford, Giger and Marsden	Waves • Transverse and longitudinal waves • Properties of waves • Wave speed • Electromagnetic waves • Refraction	Materials • Density • Fluids, viscosity and Stoke's law • Hookes Law Young Modulus Wave and Particle Nature of light	Further Mechanics • Impulse • Collisions in 2 dimensions • Circular motion and centripetal force Gravitational Fields

Autumn 2		Energy • Energy Stores and systems • Calculating Energy Change • Specific and	• Further Developments • Nuclear Decay • Alpha, Beta and Gamma Decay • Radioactive contamination • Half life Nuclear equations	• Ray Diagrams • Uses and appliances of EM waves • Radio Signals • Hazards of EM waves Magnetism • Magnetic poles and fields • Plotting fields • Electromagnetism and solenoids • Electromagnetic devices • Flemings left hand rule • Electric motors	• Wave equation • Interference and superposition • Standing waves • Radiation intensity and energy transfer • Refraction, lenses and Snell's law • Diffraction and Huygen's construction • Photoelectric effect and quantum behaviour • Energy levels in atoms • Spectra and photons • De Broglie wavelength for electrons	• Newton's gravitational law • Field strength and potential • Objects in orbit Electric and Magnetic Fields • Force fields • Radial fields and potential • Capacitors and exponential decay/growth • Magnetic forces on charges and wires • Electromagnetic induction – Faraday's law • AC electricity
Spring 1		• internal energy	Forces	Space (separates only)	Mechanics	

Spring 2		• Energy Transfers • National and global energy resources Electricity • Standard circuit Symbols • Electric Charge and Current Resistance and Potential difference	• Scalar and Vector Quantities • Contact and non-contact forces • Gravity • Resultant Forces • Vector Diagrams • Work done and energy transfer • Forces and Elasticity • Distance and Displacement • Speed • Velocity • Newton's First Law • Distance Time Graphs	• Our solar system • The formation of our solar system • Life Cycle of a star • Creating new Elements • Orbital Motion • Red Shift • The big bang Theory	• SUVAT equations • Projectiles • Momentum conservation • Potential and kinetic energy in motion • Power and work Electric circuits • Current and charge • Power and resistance • Resistivity • Potential dividers • Sensors and components • EMF and internal resistance	Nuclear and Particle Physics • Models of the atom • Particle accelerators • Creation and annihilation • The standard model, baryons and leptons Nuclear Radiation • Binding energy • Gamma absorption and half thickness • Exponential decay Oscillation • Simple harmonic motion • Pendula and springs • Resonance and damping Space • Radiation intensity and luminosity • Parallax • The HR diagram • Life cycle of stars • Doppler effect, red shift and Hubble's law
Summer 1		• Circuits and other components • Series and Parallel circuits • Power in circuits	• Acceleration • Velocity time graphs • Newton's second law • Terminal velocity • Newton's third law			

Summer2		• Direct and alternating current • Mains electricity • Danger of Mains electricity • Power and efficiency • Energy transfer in appliances • The National Grid	• Momentum • Conservation of momentum • Stopping distances • Reaction times • Factors effecting breaking distances			
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